

Testable learning outcomes for CHEM 4 – Exam #2

The slide numbers listed below correspond to those on the posted **PowerPoint slides**. For various reasons, they might not necessarily match the **recorded lectures**.

1) All CHEM 4 exams are cumulative.

- You are responsible for all the material that was covered on Exam #1
- In particular, make sure you are able to do all three types of naming: ionic, molecular, and acids (Textbook Sections 5.6 – 5.10)

2) Section 2.1 – 2.4; Significant figures and measurements (Monday, October 5)

- Know the definitions/terms on the clicker questions. [Slide 5]
- Be able to discuss and identify “exact numbers”. [Slides 6]
- Be able to discuss “measurements”. [Slides 7-8]
- Be able to make a measurement with the correct number of digits. [Slides 9, 10] and [Slide 2 from next class]
- Be able to apply the rules for identifying the number of significant figures in a measurement. [Slides 11 – 14]
- Be able to write any number in “scientific notation” in order to avoid “ambiguous zeros”. [Slide 12]
- Be able to carry out calculations with \times/\div and report the answer with the correct number of digits. [Slides 15 – 17] and [Slide 1 from next class]

3) Section 2.4 cont., 5.11; Formula mass (Wednesday, October 7)

- Know the definitions/terms on the clicker questions. [Slide 7]
- Be able to carry out calculations with $+/-$ and report the answer with the correct number of digits. [Slides 8 – 9, 11 – 13]
- Be able to calculate formula mass for a given compound and report the answer with the correct number of digits. [Slides 10 – 13] and [Slide 1 from next class]
- Not responsible for:** Chemistry in the News. [Slide 5]

4) Section 2.4 cont.; Mixed $+/-$ and \times/\div calculations (Friday, October 9)

- Be able to carry out calculations with mixed $+/-$ and \times/\div and report the answer with the correct number of digits. [Slides 4 – 8] and [Slide 1 from next class]

5) Section 2.5 – 2.6; Units/Conversions (Monday, October 12)

- Know definitions/terms on the clicker questions. [Slide 5]
- Be able to make conversion factors from equalities. [Slides 7 – 8 and 11 – 13] and [Slide 2 – 3 from next class]
- Be able to carry out metric \leftrightarrow metric, English \leftrightarrow English, and metric \leftrightarrow English conversions. [Slides 9 – 10 and 14] and [Slide 1 from next class]
- Not responsible for:** Chemistry application. [Slide 6]

6) Section 2.7 – 2.8; Units/Conversions (Wednesday, October 14)

- Be able to carry out multi-step conversions. [Slides 7 – 11] and [Slide 1 from next class]

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7) Section 2.9; Units raised to a power: Area and volume (Friday, October 16)

- a) Be able derive squared and cubed conversion factors. [Slide 6 – 8]
- b) Use squared and cubed conversion factors to carryout calculations. [Slide 9 – 12] and [Slide 1 from next class]

8) Section 2.10 – 2.11; Density (Monday, October 19)

- a) Be able to calculate density. [Slide 6 – 7]
- b) Know basic information about density. [Slide 8]
- c) Be able to use densities in calculations. [Slides 9 – 11] and [Slide 1 from next class]

9) Section 3.8 – 3.10; Energy and temperature (Wednesday, October 21)

- a) Know basic information about energy and temperature. [Slide 6 – 10]
- b) Be able to perform calculations related to energy including converting between various units of energy. [Slides 11 – 12] and [Slide 1 from next class]
- c) Be able to convert between the three temperature scales. [Not covered in lecture, but you are responsible for it]
- d) **Not responsible for:** Coronavirus Unveiled article. [Slide 4]

10) Section 3.11 – 3.12; Heat capacity (Friday, October 23)

- a) Know basic information about heat capacity. [Slide 6 – 10]
- b) Be able to perform calculations that involve heat capacity. [Slides 11 – 16] and [Slides 1 – 2 from next class]

11) Section 3.12 cont.; Heat capacity (Monday, October 26)

- a) Be able to carry out various calorimetry calculations including finding the Calorie content in food using data from a bomb calorimeter. [Slides __]